SHAILABALA WOMEN'S (AUTO.) COLLEGE, CUTTACK

SOLID STATE PHYSICS SEMESTER - V C C - XII

DEPARTMENT OF PHYSICS

SHORT ANSWER QUESTIONS

- 1. What is a crystal?
- 2. Give the difference between crystalline and amorphous materials?
- 3. Explain what do you mean by a lattice?
- 4. What is a motif or basis?
- 5. Define crystal structure.
- 6. Define unit cell and primitive unit cell.
- 7. What is a condensed matter physics?
- 8. What is a primitive cell and what is the volume in case of BCC and FCC lattice?
- 9. Name the four types of crystalline solids?
- 10. Define a metallic solid.
- 11. What is the meaning of word amorphous?
- 12. In diamond structure what is the number of nearest neighbours?
- 13. What do you mean by point groups?
- 14. What is a space group?
- 15. Give the physical significance of packing fraction.
- 16. Explain why metallic bonds are stronger than other types of bonds?
- 17. Name a particle that effects the material properties mostly.
- 18. What is the mean distance between atoms.
- 19. In which bond the term electron sea exist.
- 20. What is the atomic diameter of BCC lattice?
- 21. What are the Miller indices of octahedral plane in cubic crystal?
- 22. Define translational operators.
- 23. How a Wigner-Seitz cell is drawn.
- 24. Which type of structure NaCl possess?
- 25. Why y-rays are not used to study crystal structure?
- 26. What is the difference between direct lattice and reciprocal lattice?
- 27. What is first Brillouin Zone?
- 28. Define reciprocal lattice. How is it constructed?
- 29. What is the need of using reciprocal lattice? Why should we not use direct lattice.
- 30. Why do crystals diffract X-rays?
- 31. Define atomic scattering factor.
- 32. What are the considerations in deriving Laue's equations?
- 33. What is the significance of Brillouin Zone?
- 34. What is meant by axis of symmetry?
- 35. What is the coordination number in case of BCC structure?
- 36. Find the packing structure of SC structure.
- 37. Define a photon.
- 38. Which crystals exhibit optical phonon modes?

- 39. Which restriction on the k-values for the lattice vibrations are found for one dimensional chain with one and two atoms per cell respectively?
- 40. Is the group velocity same for optical and acoustical lattice vibration? Explain.
- 41. What is common between phonon and a photon?
- 42. How does elastic vibrations differ from electromagnetic waves?
- 43. Which statistics does phonon obey?
- 44. At what temperatures the result of Dulong-Petit's law is in agreement with the experimental values?
- 45. Plot a graph to show the general temperature dependence of specific heats of solids?
- 46. Give the expression for the average energy for one dimensional oscillator according to classical theory of specific heats?
- 47. When does the expression for specific heat of solid in Debye's theory approach the classical value 3R?
- 48. What do you understand by the term magnetization?
- 49. Distinguish between para, dia and ferromagnetic materials
- 50. Explain the classification of magnetic materials.
- 51. What are domains?
- 52. Discuss domain theory of ferromagnetism.
- 53. What is Weiss theory of ferromagnetic materials?
- 54. What is Curie Law?
- 55. What is Curie temperature?
- 56. What is hysteresis curve?
- 57. What is hyteresis loss?
- 58. Write short notes on
- 59. (i) Classification of magnetic materials.
- 60. (ii) Weiss theory of ferromagnetism. N m
- 61. Calculate magnetic moment of an atomic dipole.
- 62. Explain Langevin's theory of paramagnetism
- 63. Explain Langevin's theory of diamagnetism.
- 64. Define dielectric constant of a material.
- 65. Define dielectric strength.
- 66. What do you mean by dielectric strength?
- 67. What do you mean by polar and non-polar molecules? Give two examples each.
- 68. What are polar and non-polar dielectrics?
- 69. Define the terms dipole moment and polarization.
- 70. What do you mean by polarization of a dielectric?
- 71. Define two electric vectors D and E, What is the relation between them?
- 72. Define electric polarization vector P . What is its S.I. units?
- 73. Explain the term permittivity

- 74. What do you understand by electric displacement, susceptibility and permittivity?
- 75. What is dielectric polarization and electric displacement vector?
- 76. Name mechanism of dielectric polarization.
- 77. Derive the relation between dielectric constant and suscepitibility.
- 78. Define atomic polarizability.
- 79. Name different types of polarization.
- 80. Explain that electric field inside the dielectric is reduced when it is placed inside the electric field.
- 81. What do you understand by local field at an atom?
- 82. What is local field at an atom?
- 83. What do you mean by the term 'LASER'?
- 84. What is basic principle of LASER?
- 85. What is population inversion and why it is important for lasing?
- 86. Why it is easy to use two-level atom in laser?
- 87. Are there any two level atom laser? What are they
- 88. What do you mean by stimulated absorption of light?
- 89. What is stimulated emission of light?
- 90. What is the need of stimulated emission of light?
- 91. What do you mean by the term population inversion?
- 92. What do you mean by pumping?
- 93. What do you mean by optical pumping?
- 94. What is difference between the continuous and pulsed laser output?
- 95. Who made the first Ruby laser? What is its output wavelength?

LONG QUESTIONS

- 1. Discuss the quantization of lattice vibrations? Explain the concepts of phonon? Do
- 2. Describe the lattice vibrations of monoatomic linear lattice and obtain an expression for the dispersion relation for lattice vibrations of monoatomic linear chain?
- 3. Obtain the various vibrational modes of a linear monoatomic lattice.
- 4. Derive an expression for the energy and momentum of phonons and explain normal process and umklap process
- 5. Obtain an expression using the inelastic scattering of photons by phonons, for the
- 6. frequency of phonon emitted in the process.
- 7. What is a phonon? Describe the vibrational modes of a diatomic linear lattice? Name the different branches of the dispersion relation curve? What is the difference between the two branches?
- 8. Define molar specific heat at constant volume and lattice heat capacity? Explain dulong and Petit's law. What are the limitations of this law?
- 9. What are para, ferro and diamagnetic substances? Give their properties.
- 10. Explain Atomic Theory of Magnetism (Qualitative).
- 11. Discuss Langevin's theory of Diamagnetism and obtain the expression for diamagnetic susceptibility.
- 12. Discuss Langevin's theory of paramagnetism and derive and expression for paramagnetic susceptibility.
- 13. Discuss Domain theory of Ferromagnetism .
- 14. Explain diagmanetism, paramagnetism and ferromagnetism on the basis of dipoles of atoms.
- 15. Discuss the Weiss theory of ferromagnetism. Show from Langevin function that ferromagnets lose their spontaneous magnetization above Curie temperature.
- 16. Explain Curie law and Curie temperature.
- 17. Give Weiss Theory of Ferromagnetism.
- 18. Give an expression for paramagnetic susceptibility. How does the paramagnetic susceptibility of a material vary with temperature?
- 19. Draw the B-H curve for a ferromagnetic material and identify the retentivity and coercivity on the curve.
- 20. Explain ferromagnetism on the basis of Domain theory
- 21. Explain the use of hysteresis curve.
- 22. Explain what do you understand by Hysteresis, Retentivity and Coercivity. How will you determine the value of retentivity and coercivity from a loop?
- 23. State Curie's law. What is Curie temperature?
- 24. Explain the term hysteresis and prove that hysteresis loss per cycle of magnetization is equal to the area of the B-H loop.
- 25.20. Write notes on:

- Electron spin and paramagnetism.
- Domain theory of ferromagnetism. (ii)
- (iii) **Ferrites**
- (iv) Diamagnetism
- 26. What are ferrites? How do they differ from ferromagnetic substances?
- 27. What is dielectrics? Define dielectric constant of a material.
- 28. Explain the phenomenon of dielectric polarization in dielectric materials. Define dielectric constant.
- 29. Explain the terms atomic polarizability, dielectric susceptibility and atomic dipole moment.
- 30. Define local field at an atom. Derive the relation between local field and polarization of a dielectric.
- 31. Derive Clausius-Mossotti relation between polarizability and dielectric constant of a solid.
- 32. Derive Clausius-Mossotti relation
- 33. Derive molecular interpretation of Clausius-Mossotti equation?
- 34. Deduce the Clausius-Mossotti relation for polarization of a medium and explain how it could be used to determine dipole moment of polar molecules.
- 35. Explain classical Theory of electronic polarizability.
- 36. Discuss different types of polarization. Discuss the frequency dependence of these polarizations.
- 37. Define polar and non-polar molecules. Deduce Clausius Mossotti relation for nonpolar dielectrics.
- 38. What are the properties of laser light?
- 39. Write four properties of laser light.
- 40. What do you mean by spatial coherence and temporal coherence? Explain
- 41. Explain the mechanism of production of laser beam.
- 42. Explain the following terms: (i) Spontaneous emission (ji) Stimulated emission (iii) Population inversion (iv) Metastable state
- 43. What are the characteristics that distinguish laser from ordinary source of light?
- 44. Explain directionality and monochromaticity for laser light,
- 45. Distinguish between spontaneous and stimulated emissions. Which one of the two predominates in optical region at room temperature?
- 46. Discuss in details the concept of directionality, monochromaticity, intensity and coherence of laser light.
- 47. What do you understand by spatial and temporal coherence of light source? What is coherence time and coherence length?

- 48. Define coherence, temporal coherence, spatial coherence, coherence time and
- 49. Discuss the periodicity character of potential in a crystal. State Bloch theorem in this
- 50. Discuss Kroning-Penny model. Using this model show that the energy spectrum of electron consists of a number of allowed energy bands separated by forbidden regions.
- 51. What do you mean by effective mass? Deduce the relation for effective mass of an electron. Show how it is different from rest mass of the electron.
- 52. Differentiate n-type and p-type semiconductors. How does the conductivity and mobility of electrons and holes change with temperature?
- 53. Obtain an expression for density of electrons in conduction bond of n-type semiconductor.
- 54. Obtain an expression for density of holes in conduction band of p-type semiconductor.
- 55. Discuss the motion of an electron in one dimension according to band theory and show how the energy, velocity and effective mass vary as function of wave vector.
- 56. Explain Hall effect? Derive an expression in for the Hall coefficient of semiconductor on two band model of carriers. Discuss Four probe method and uses of Hall Effect.
- 57. What is superconductivity? Discuss the temperature dependence of resistivity in superconducting material
- 58. Define superconductors. What are the properties of superconductors?
- 59. What is superconductivity? What are the properties which do not change in super.
- 60. Define superconductors. What are the properties which change in superconducting transition?